

Fission Weapons

Diffusion Equations run amuck

By Scott Stafford



The Desired Result

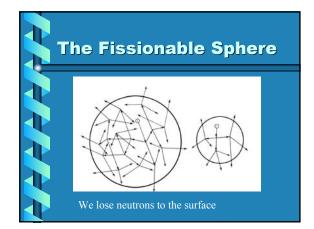
 The process of fissioning an atom produces, on average, 170 MeV per fission event. This is approximately 10° times the heat of reaction per atom in an ordinary combustion process.
 The desire was to harness this energy into an explosive effect.

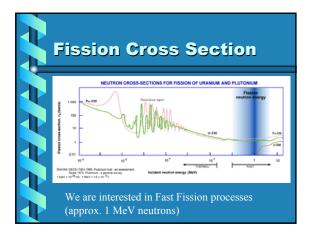
 In a mass of 1 kg of U-235 with 2.58x10²⁴ nuclei it would require n=81 generations for all U-235 atoms to fission(assuming no loss to the surface). Since each fission occurs in about 10⁴ seconds, the 81 generations would occur in 0.81 microseconds.

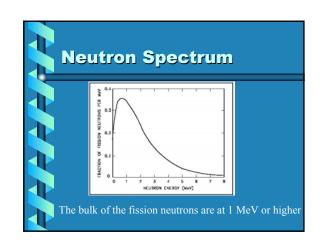
The Desired Result

 During this entire fissioning process, the energy released is making the material very hot and developing tremendous pressure...this is the explosive result. If the reaction were to proceed at 10% efficiency the Uranium would be initially heated to approx. 10¹⁰ °C in less than a millionth of a second. The pressure rises similarly.

But there are problems....

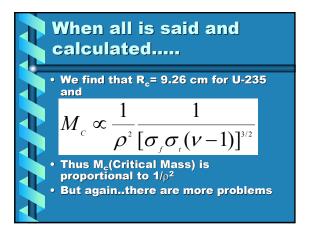


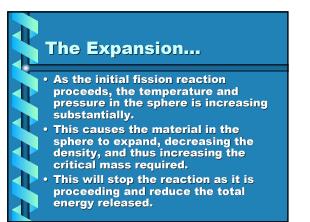


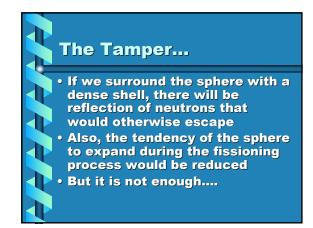


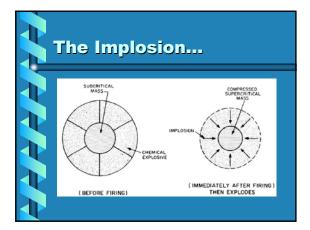
Diffusion Equation
Neutrons Produced per
Neutrons los to next fission

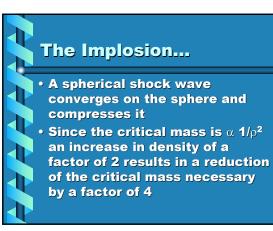
$$dN = -div(j) + \frac{v-1}{r}$$
 Neutrons los to next fission
 $dN = -div(j) + \frac{v-1}{r}$ Neutrons los to next fission
 $j = -DgradN$
Mean time between fissions
and....
 $dN = D\Delta N + \frac{(v-1)}{\tau} N$











The Implosion....

- This will cause the previous subcritical mass to go "prompt" super critical.
- Also, the implosion helps initially reduce the sphere's tendency to expand near the final neutron generations.
- Thus the efficiency of the energy released is increased (≈20%)