Morti Student Name: ID: Page 1 of 3 123456 **ENGINEERING PHYSICS 4D3/6D3** DAY CLASS Dr. Wm. Garland **DURATION: 50 minutes** McMASTER UNIVERSITY MIDTERM EXAMINATION October 18, 2005 **Special Instructions:** 1. Closed Book. All calculators and up to 6 single sided 8 ½" by 11" crib sheets are permitted. 2. Do all questions. 3. The value of each question is as indicated. TOTAL Value: 100 marks THIS EXAMINATION PAPER INCLUDES 3 PAGES AND 3 QUESTIONS. YOU ARE RESPONSIBLE FOR ENSURING THAT YOUR COPY OF THE PAPER IS COMPLETE. 1. [40 marks] Consider a cubic homogeneous reactor of dimension a x a x a in an ocean of water. For a 1-group neutron diffusion model in steady state: State the relevant one group neutron balance equations for the reactor and the ocean. Sketch the situation to clarify notation and layout. 12 ) + (YZF - Zac ) \$ b. State the required boundary conditions for an analytical solution. 12 B.C's required (+ pat (0,0,0) to set amplitude) x= 0/2 @ also for y + 2 } 3 B.C.s. = Jx = -D34 c. If you were solving this numerically, what boundary conditions would you use? Just set \$ = 0, some for xxy, where max e edged) the numerical grid. This is consistent with perence equation of d. Why are the B.C.s different for the numerical and analytical cases?

2. [30 marks] A bare cubic reactor can be characterized by one group neutron diffusion. The reactor is composed of a lattice of fuel channels with D<sub>2</sub>O coolant surrounded by D<sub>2</sub>O moderator like in a CANDU reactor, but we will model it as a homogeneous mixture. For this setup assume D = 6.0 cm.,  $\Sigma_a = 6.0 \times 10^{-4}$  cm.<sup>-1</sup>, critical height =

width = length = 100 cm.  $K=1=\frac{2}{1+\frac{2}{B^2L^2}}$  Since critical. (samething)

Bz = 3(11/102)z = 0.0054PI

L2 = D/E = 1 x10+ cm2

 $L^{2} = D/\xi_{a} = 1 \times 10^{2}$   $P(\xi_{a}) = (1 + B^{2}L^{2}) \cdot \xi_{a} = 0.01831$   $= (1.837 \times 10^{2} \text{ cm})$   $= 7 \xi_{f}$ 

b. You, as reactor designer, are exploring alternate designs. If you replace the D<sub>2</sub>O coolant with  $H_2O$ , the material properties change to D = 0.3 cm.,  $\Sigma_a = 0.03$  cm.<sup>-1</sup>. What would be the new dimensions, keeping the same fuel and keeping it cubic in shape?

 $B^{2} = \frac{\sqrt{26}}{50} - 1 = \frac{\sqrt{1.0837} \times 10^{2}}{\sqrt{3}} - 1$ 2017

Uh-sh! Laskolike we cannot make the reac entical with the same fuel if we use H2O, no matter what the dimensions.

in must increase the amount of fuel. This is what happened for the new CANDU design (ACR) where switching to H2O coolant forced the use of enriched fuel?

[ Note: the values used in this problem are begue and are not CANDU numbers ]

3. [30 marks] Consider a processing facility for the production of an isotope B, consisting of a small container inside a reactor. The container is small enough and the absorption cross sections low enough that the neutron flux is uniform in the container. The isotope B is produced by an  $A(n, \gamma)B$  and B decays to C with a half

life,  $T_{1/2}$ ; there are no other significant nuclear reactions in the container. Isotopes A,  $S_{6,0}$   $S_{C,0}$  B and C can be added and removed from the container at will.

a. State the governing balance equations for A, B and C.

 $\frac{\partial N_B}{\partial t}(t) = S_B(t) + S_A N_A D_B - \lambda_B N_B(t)$   $\frac{\partial N_C}{\partial t}(t) = S_C(t) + \lambda_B N_B(t) \left(\lambda_B = \frac{2n^2}{T_L^2}\right)$ 

b. Propose a numerical solution algorithm, showing the finite difference equations and a simple flow diagram showing how the algorithm works.

NA = NA + IST (SA-JANA \$)

Let's use implicit:

=> NATHAT = NATH ATSA

 $\sim N_B^{t+\Delta t} = \frac{N_B^t + \Delta t S_B + \nabla_A N_A^t \phi}{(1 + \Delta t \lambda_B)}$ 

~ Nettat = Net + At (SB + 2BNB)

Start)-set I.C: NA= NAO, etc.

No trat

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D, SA, SB, Sc are all import Values. In general, they could be