

1. [Serway Chapter 22 Problem 2, pg 640]
A heat engine performs 200 J of work in each cycle and has an efficiency of 30%. For each cycle, how much thermal energy is
 - a. absorbed and
 - b. expelled?

2. [Serway Chapter 22 Problem 8, pg 640]
A heat engine operates between two reservoirs at 20°C and 300°C. What is the maximum efficiency possible for this engine?

3. [Serway Chapter 22 Problem 12, pg 640]
A heat engine operates in a Carnot cycle between 80°C and 350°C. It absorbs 2.0×10^4 J of thermal energy per cycle from the hot reservoir. The duration of each cycle is 1.0 s.
 - a. What is the maximum power output of this engine?
 - b. How much thermal energy does it expel in each cycle?

4. [Serway Chapter 22 Problem 15, pg 641]
The efficiency of a 1000 MW nuclear power plant is 33%; that is, 2000 MW of heat is rejected to the environment for every 1000 MW of electrical energy produced. If a river of flow rate 10^6 kg/s were used to transport the excess thermal energy away, what would be the average temperature increase of the river?

5. [Serway Chapter 22 Problem 20, pg 641]
A gasoline engine has a compression ratio of 6 and uses a gas for which $\gamma = 1.4$.
 - a. What is the efficiency of the engine if it operates in an idealized Otto cycle?
 - b. If the actual efficiency is 15%, what fraction of the fuel is wasted as a result of friction and unavoidable heat losses? (Assume complete combustion of the air-fuel mixture.)

6. [Serway Chapter 22 Problem 25, pg 641]
A refrigerator has a coefficient of performance equal to 5. If the refrigerator absorbs 120 J of thermal energy from a cold reservoir in each cycle, find
 - a. the work done in each cycle and
 - b. the thermal energy expelled to the hot reservoir.