Nuclear Training Course PI27 TIMS Ref. PI2007

Nuclear Theory

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NUCLEAR TRAINING DEPARTMENT COURSE PI 27 NUCLEAR THEORY

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NUCLEAR TRAINING COURSE

COURSE PI 27

NUCLEAR THEORY

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Nuclear Theory - Course PI 27

OBJECTIVES

At the conclusion of this course the trainee will be able to:

427.00-2 Radioactivity

- 1. For α , β , γ decays
 - (a) Write typical equations for each.
 - (b) List the physical properties.
 - (c) Discuss interactions with materials.
- 2. Know how to shield against alphas and betas.
- 3. Know how to shield against γ rays and be able to calculate γ ray shielding of 1/2 value layers.

227.00-1 Nuclear Structure

- 1. Explain the concept of binding energy.
- 2. Discuss the stability of nuclei in terms of their neutron-proton ratio.
- 3. From a plot of n against p say what emission a given nuclide is likely to undergo.
- 4. Be able to follow a decay chain from a radioactive nuclide until a stable nuclide is reached.
- 5. Define the unit of activity, the Becquerel.
- 6. State the basic law governing radioactive decay.
- 7. State the relationship between decay constant (λ) and half life ($t_{\frac{1}{2}}$).

227.00-2 Neutron Reactions

- 1. Differentiate between elastic and inelastic collisions.
- Explain the importance of elastic collisions to the operation of CANDU reactors.

- 3. State the name of the four types of inelastic collisions giving an example of each $(7A^A)$ type example is acceptable).
- 4. Differentiate between spontaneous and induced fission.
- 5. Explain a self sustaining chain reaction.
- 6. Write the equations for the formation of $_{94}$ Pu²³⁹ in our reactors.
- 7. Define:
 - (a) Prompt Neutrons
 - (b) Delayed Neutrons
 - (c) Delayed Neutron Precursors
 - (d) β Delayed Neutron fraction
 - (e) υ Neutrons Emitted per Fission
 - (f) Photoneutron
 - (g) Fast neutrons
 - (h) Thermal neutrons
- 8. Give the distribution of energy released by the fission of U-235.

227.00-3 Neutron Cross Sections, Neutron Density and Neutron Flux

- 1. Define:
 - (a) Microscopic Neutron Cross Section and the units.
 - (b) Macroscopic Neutron Cross Section and the units.
 - (c) Neutron Density and the units.
 - (d) Neutron Flux and the units.
- 2. Relate σ_a , σ_f and $\sigma_{n,\gamma}$.
- 3. Discuss how the microscopic cross sections of U-238 and U-235 vary with neutron energy.
- 4. Write reaction rates.
- 5. Be able to extract data from the chart of the nuclides.

227.00-4 Thermal Reactors

- 1. Discuss the properties of a moderator including the number of collisions required to thermalize a neutron, scattering cross section, and absorption cross section.
- 2. Define the moderating ratio.
- 3. Explain the practical significance of the fact that D₂O, compared to H₂O has a lower scattering cross section and requires more collisions to thermalize a neutron.
- 4. Discuss the effect of downgrading the moderator or heat transport fluid.
- 5. Define lattice pitch.
- 6. Explain what "over moderated" means and why Hydro's reactor are over moderated.
- 7. Explain why increasing or decreasing the lattice pitch from its optimum value causes reactivity to change.

227.00-5 Neutron Multiplication Constant and Reactivity

- 1. Define k both in words and in terms of the six factors.
- 2. State when the word definition is not valid.
- 3. Define and explain each of the six factors in k.
- 4. Sketch a neutron life cycle using the six factors.
- 6. Define:
 - (a) Critical
 - (b) Subcritical
 - (c) Supercritical
- 7. State and explain the significance of the four-factor formula for k_{∞} .
- 8. Define and calculate values of reactivity and of reactivity worths.
- 9. Calculate values of the six factors given a neutron life cycle.

227.00-6 Neutron Flux Distribution

- 1. Discuss the functions of a reflector.
- 2. Discuss the effects of a reflector.
- 3. Explain why flux flattening is desirable.
- 4. Discuss the four methods of flux flattening used.
- 5. Sketch the flux shapes showing the effect of each of the flux flattening methods.
- 6. Discuss the effect of reactor size and shape on neutron leakage.

227.00-7 Effect of Fuel Burnup

- 1. State and explain the units used for fuel burnup.
- 2. Explain why the combined reactivity worth due to U-235 and Pu-239 initially increases then decreases with burnup.
- 3. Explain how and why each of the four factors of k_{∞} changes with fuel burnup.
- 4. Explain how and why the delayed neutron fraction (β) changes with fuel burnup.

227.00-8 Changes in Reactor Power with Time

- 1. Physically explain the effect of delayed neutrons on changes in reactor power.
- 2. Given the formula, $P(t) = \frac{\beta}{\beta \Delta k}$ po e $\frac{\lambda \Delta k}{\beta \Delta k}$, solve calculational type problems.
- 3. Explain the concept of the prompt jump.
- 4. Define prompt criticality and explain why it is undesirable. Explain its dependence upon fuel composition and fuel burnup.

227.00-9 Source Neutron Effects

1. State the sources of neutrons and their approximate magnitudes.

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- 2. State and use the formula $S_{\infty} = \frac{So}{1-k}$.
- 3. Define and explain the significance of the subcritical multiplication factor.
- 4. Calculate k in a subcritical reactor given appropriate data.
- 5. State that, for a sub-critical reactor, the closer k is to one, the longer it takes for power to stabilize after a reactivity change.

227.00-10 Power and Power Measurement

- 1. Explain how thermal power is measured.
- 2. Explain why neutron power must be calibrated to thermal power.
- 3. Explain the reasons why neutron power is used for control and protection of the reactor.
- 4. State the relationship between reactor period and rate log N. (For engineers: Prove the relationship).
- 5. Make an accurate sketch of the rundown of neutron power after a trip justifying times and power levels used.
- 6. Discuss the rundown of thermal power after shutdown.

227.00-11 Fission Product Poisoning

- 1. Explain how xenon and iodine are produced in the reactor and how they are lost from the reactor.
- 2. Write the differential equations for the concentration of xenon and iodine and define each term.
- 3. State the magnitude of the production and loss terms for xenon at equilibrium in our larger reactors.
- 4. Define Xenon Load and Iodine Load.
- 5. Explain what Xenon Simulation is.
- 6. Sketch and explain the behaviour of xenon after a trip from full power.
- 7. State and explain the two conditions necessary for a Xenon Oscillation.

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- 8. Explain what a Xenon Oscillation is and how one may be started.
- 9. Explain why samarium growth after shutdown is not a problem.

227.00-12 Reactivity Effects Due to Temperature Changes

- 1. Explain why a negative fuel temperature coefficient of reactivity is desirable.
- Give two undesirable effects of having a negative fuel coefficient.
- 3. Explain why the fuel temperature coefficient is more important than either the coolant or moderator temperature coefficient.
- 4. Explain why the fuel temperature coefficient is negative and why its value changes from fresh to equilibrium fuel.
- 5. Define the power coefficient and give a typical value.
- 6. Define the void coefficient.

227.00-13 Reactivity Control

- 1. List the various in-core reactivity worth changes, typical magnitudes of the changes, and the time period over which the changes occur.
- 2. Discuss general methods of reactivity control in terms of their effect on the six factors of k.
- 3. Given a specific method of reactivity control (eg, Moderator Level Control) discuss its advantages and disadvantages.
- 4. List and discuss the advantages and disadvantages of each of the presently used shutdown systems.

227.00-14 The Approach to Critical

- 1. Explain why the initial approach to criticality is potentially hazardous.
- 2. Explain how inverse count rate is used to predict the critical value of the controlling reactivity mechanism.

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